

## PHOTOCATALYTIC PROPERTIES OF FERROCYANIDES

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### ABSTRACT

A number of heavy metal hexacyanides are tested for their ability to catalyse photo-oxidation of water with  $\text{Ru}^{2+}$  as the sensitizer and  $\text{K}_2\text{S}_2\text{O}_8$  as the sacrificial agent. Strongest catalytic activity is seen in  $\text{Zn}_3(\text{Fe}(\text{CN})_6)_2$ ,  $\text{Cd}_3(\text{Fe}(\text{CN})_6)_2$  and  $\text{Fe}_3(\text{Fe}(\text{CN})_6)_3$ . An aqueous suspension of cupric Ferrocyanide behaves differently photogenerating  $\text{H}_2$  and  $\text{O}_2$  with visible light under sacrificial conditions in the absence of electron or hole transfer agent. Semiconducting properties of Ferronoides and their relevance to catalytic activity are discussed.