

ICMAT 2013

7TH INTERNATIONAL CONFERENCE ON
MATERIALS FOR ADVANCED TECHNOLOGIES

30 JUNE - 5 JULY, SUNTEC SINGAPORE

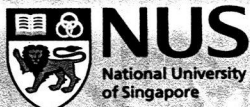
TECHNICAL PROGRAMME



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15:00 A2.2-3
Thermodynamics Study of Nano-size Lithium Cobalt Oxide Cathode Material
Shahnaz GHASEMI^{1*}, Kenza MAHER¹, Harry HOSTER¹, Rachid YAZAMI²
¹TUM CREATE, Singapore, ²School of Materials Science and Engineering, Nanyang Technological University, Singapore

15:15 A2.2-4
The Synthesis, Structural Characterisation and Interlayer Mixing of Layered LiNi_{0.4}Mn_{0.4}Co_{0.2}O₂ Cathode Material
Tze Qing TAN^{1*}, Mohd Sobri IDRIS¹, Azmi RAHMAT¹, Shamsul Baharin JAMALUDIN¹, Rozana Aina MAULAT OSMAN¹, Zul Azhar ZAHID JAMAL¹
¹Sustainable Engineering Research Cluster, School of Materials Engineering, University Malaysia Perlis, Malaysia

15:30 A2.2-5
General Template-free Synthesis of the Mixed Metal Oxide Complex Hollow Microspheres and Their Lithium Storage Properties
Genqiang ZHANG^{1*}, Harry HOSTER¹, David LOU²
¹TUM CREATE, Singapore, ²School of Chemical and Biomedical Engineering, Nanyang Technological University, Singapore

15:45 A2.2-6
Development of Cathode Materials for Lithium Ion Rechargeable Batteries Based on the System Li(ni^{1/3} Mn^{1/3} Co^{1/3}-xmx)₂, M = Cu, Zn, Fe, Mg ; X = 0 to 0.33
Athula WIJAYASINGHE^{1*}, Pushpaka SAMARASINGHE², Gayani AMARAWEEERA^{1*}, Manoj WIJESINGHE¹, Lakshman DISSANAYAKE¹
¹Institute of Fundamental Studies, Sri Lanka, ²Department of Chemistry, University of Oslo, Norway

A3.1: Anodes II

Mon - 1 Jul 13 | 16:30 - 18:00 | Room 326

Session Chair(s): Jaephil Cho & Dominik Samuelis

16:30 A3.1-1
A High Capacity Molybdenum Oxide-Reduced Graphene Oxide Composite as Anode Material for Li Ion Battery
wei TANG^{1*}, Kian Ping LOH^{2*}, M. V. REDDY³
¹Chemistry, National University of Singapore, Singapore, ²National University of Singapore, Singapore, ³Department of Physics, Solid State Ionics/Advanced Batteries Lab, Singapore

16:45 A3.1-2
Carbon-encapsulation of F-doped Li₄Ti₅O₁₂ as High Performance Anode Material for Reversible Lithium Ion Storage
Yue MA^{1*}, Jim Yang LEE^{2*}
¹NUS Graduate School for Integrative Sciences and Engineering, National University of Singapore, Singapore, ²National University of Singapore, Singapore

17:00 A3.1-3
Three-dimensional ZnCo₂O₄/C Core/shell Nanowire Array Anode for Lithium-ion Batteries
Qinqin XIONG^{1*}, Jiangping TU^{2*}, Chang Dong GU³, Xiuli WANG²
¹Zhejiang University, China, ²Department of Materials Science and Engineering, Zhejiang University, China, ³Department of Materials Science and Engineering, China

17:15 A3.1-4
Free Standing Multi-walled Carbon Nanotubes Based Anode for Lithium Ion Battery
Indu ELIZABETH^{1*}
¹Physics and engineering of carbon, National Physical Laboratory, India

17:30 A3.1-5
Synthesis of Reduced Graphite Oxide (RGO) - Fe₃O₄ Composite with the Aid of Microwave and Electrochemical Characterization for Lithium Ion Battery Anode
Ravikumar RAMAN^{1*}
¹ECPS, CSIR-CECRI, India

17:45 A3.1-6
Hierarchical Fe₃O₄-cuo Hybrids as Lithium-ion Battery Anodes
Somaye SAADAT^{1*}, Qingyu YAN^{1*}, Rachid YAZAMI²
¹Nanyang Technological University, Singapore, ²School of Materials Science and Engineering, Nanyang Technological University, Singapore

A3.2: Redox Molecular Grafting / Electrolytes I

Mon - 1 Jul 13 | 16:30 - 18:00 | Room 325

Session Chair(s): Erik Jämstorp Berg & R. Prasada Rao

16:30 A3.2-1 (Invited)
Interest of Redox Molecular Grafting for Energy Storage Applications
Lenaic MADEC¹, Bernard LESTRIEZ¹, Kalid SEID², Jean-Claude BADOT³, Bernard HUMBERT¹, Philippe MOREAU¹, Dominique GUYOMARD^{4*}, Joel GAUBICHER¹
¹Institute of Materials Jean Rouxel (IMN), France, ²Supelec, France, ³Chimie ParisTech, France, ⁴Electrochemical Energy Storage, Institute of Materials Jean Rouxel (IMN) - CNRS, France

17:00 A3.2-2 (Invited)
Solid-like Electrolytes for Modern Rechargeable Batteries
Aninda BHATTACHARYYA^{1*}
¹Solid State and Structural Chemistry Unit, Indian Institute of Science, India

17:30 A3.2-3
Electrical, Optical and Structural investigations of Electrolyte Films - Pure and MnSO₄ Doped PVA/PVP Polymer Blend
Sudha KAMATH^{1*}, Chandramani R², Radhakrishna MC³
¹Department of Physics, R V College of Engineering, India, ²Physics, Dayananda College of Engineering, India, ³Physics, Bangalore University, India

17:45 A3.2-4
Bio-inspired Lithium Salt Free Polymer Electrolyte Membranes
R. Shanti RAJANTHARAN^{1*}, Ramesh SUBRAMANIAM¹, Ramesh KASI¹
¹Department of Physics, University of Malaya, Malaysia

18:00 A3.2-5
Functionized Nano-porous Polymer Separator-cum-electrolyte for Lithium-ion Batteries
Rupesh ROHAN^{1*}, Cheng HANSONG^{1*}, Sun YUBAO¹
¹Department of Chemistry, National University of Singapore, Singapore

Development of Cathodes for Lithium Ion Rechargeable Batteries Based on the System $\text{Li}(\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3-x}\text{M}_x)\text{O}_2$, $\text{M} = \text{Cu, Zn, Fe, Mg}$; $x = 0$ to 0.33

Athula Wijayasinghe¹, Pushpaka Samrasinghe², Manoj Wijesinghe¹, Kobiga Sivabalasatkunam¹, Gayani Amaraweera¹ and Laxhman Dissanayake¹

¹*Institute of Fundamental Studies, Hantana Road, Kandy, Sri Lanka,*

²*Department of Chemistry, Center for Materials Science and Nano Technology University of Oslo University, Oslo, Norway*

NMC compositions such as $\text{Li}[\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3}]\text{O}_2$ are presently being commercialized as cathode materials for rechargeable Li-ion batteries (LIB). However, with very expensive cobalt content in the cathode, batteries having NMC compounds are limited to consumer electronics and still too costly for vehicular applications. Therefore considerable efforts are being made by different research groups worldwide to reduce the material cost by fully or partially substituting expensive Co with other cheaper elements. Accordingly, this study was based on developing less expensive but performance enhanced cathode materials for the LIB based on the system, $\text{Li}(\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3-x}\text{M}_x)\text{O}_2$, $\text{M} = \text{Mg, Fe, Zn, Cu}$ and $x = 0$ to 0.33 .

For this study, material synthesis was done by Pechini and Glycine Nitrate combustion techniques, followed by calcining at $800\text{-}1000^\circ\text{C}$. X-ray diffractometry and Scanning Electron Microscopy were used for phase analysis and particle characterization, respectively. Dense pellets sintered at 1000°C were used for electrical characterization by d.c. four probe technique. Bellcore-type electrodes were fabricated via tape casting and their electrical conductivity was determined by the van der Pauw technique. Electrochemical testing was performed in an argon filled glove box with $\text{Li}/\text{LiPF}_6/\text{Li}(\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3-x}\text{M}_x)\text{O}_2$ coin cells cycling between $3.0 - 4.5$ V.

The phase analysis revealed the formation of solid solutions of appropriate layered $\text{Li}(\text{Ni}_{1/3}\text{Co}_{1/3}\text{Mn}_{1/3})\text{O}_2$ phase of R3m structure in the prepared compositions up to $x = 0.11$, except with Mg. Further, most of these substituted compositions showed considerably higher electrical conductivity than the base material $\text{Li}(\text{Ni}_{1/3}\text{Co}_{1/3}\text{Mn}_{1/3})\text{O}_2$ (where it was 0.9×10^{-5} S/cm). In the cell studies, the Fe substituted composition with $x = 0.11$, showed a specific capacity of 122 mAhg^{-1} , which is comparable to the specific capacity of the state of the art LiCoO_2 cathode material of LIB. Hence this study revealed the possibility of preparing cheaper cathode materials suitable for LIBs.