

CONDUCTIVITIES AND PHASE RELATIONSHIP OF
 $\text{Na}_{1+x}\text{Zr}_2\text{Si}_x\text{V}_y\text{P}_{3-x-y}\text{O}_{12}$ FAST ION CONDUCTORS

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$\text{Na}_{1+x}\text{Zr}_2\text{Si}_x\text{V}_y\text{P}_{3-x-y}\text{O}_{12}$ fast ion conductors were synthesized from mixtures of $\text{Na}_3\text{PO}_4\text{-ZrO}_2\text{-SiO}_2\text{-V}_2\text{O}_5$ in platinum vessel at temperatures of 900 - 1200°C. Like $\text{Na}_{1+x}\text{Zr}_2\text{Si}_x\text{P}_{3-x}\text{O}_{12}$ system and $\text{Na}_{1+x}\text{Hf}_2\text{Si}_x\text{P}_{3-x}\text{O}_{12}$ system, the R3C and C2/sc solid solution can be formed in the $\text{Na}_{1+x}\text{Zr}_2\text{Si}_x\text{V}_y\text{P}_{3-x-y}\text{O}_{12}$ system with a wide composition range. The phase relationship was identified by X-ray diffraction. The limit of composition range for single V-NASICON phase has been explored and will be presented.

The electrical conductivities and activation energies of $\text{Na}_{1+x}\text{Zr}_2\text{Si}_x\text{V}_y\text{P}_{3-x-y}\text{O}_{12}$ system were measured by a.c. impedance technique on sintered discs from room temperature to 400°C. The maximum conductivity is $0.66 \times 10^{-1} (\text{cm})^{-1}$ at 350°C and the activation energy is 25.85 KJ/mole. Some other properties will be presented and correlated with composition change.